Microbial Transformation of Pyrethrosin

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Microbial transformation of the germacranolide pyrethrosin (1) using *Rhizopus nigricans* NRRL 1477 has resulted in the isolation of 6α -acetoxy- 1β , 4α -dihydroxy-5, 7α *H*, 8β *H*-eudesm- 11β ,13-dihydro-8,12-olide (5), a new eudesmanolide-type metabolite, in addition to the previously reported eudesmanolides: **2**, **3**, **4**, and **6**. The structure elucidation of these metabolites was based primarily on 1D and 2D NMR analyses. The isolated metabolites exhibited cytotoxic, antifungal, and antiprotozoal activities.

Pyrethrosin (1), the first recognized germacrane sesquiterpene lactone,^{1,2} was isolated from the flowering heads of *Chrysanthemum cinerariaefolium* Visiani³ (Asteraceae). Pyrethrosin exhibits diverse biological activities including cytotoxicity,⁴ antibacterial,⁵ antifungal,⁶ molluscicidal,⁷ and phytotoxicity.⁸

Microbial metabolism studies have been used as inexpensive in vitro model systems to predict mammalian metabolism or to increase the efficacy of a drug through metabolic activation.⁹ Pyrethrosin (1), with its diverse bioactivity range, was subjected to microbial metabolism in an attempt to find more active analogues. Twenty-four growing cultures were screened for their ability to biotransform 1. Of these, *Cunninghamella elegans* NRRL 1392 and *Rhizopus nigricans* NRRL 1477 were able to completely deplete and transform 1 into the same five more polar metabolites. *R. nigricans* (NRRL 1477) was chosen for preparative-scale fermentation due to its higher metabolite yield as compared to *C. elegans* NRRL. Repeated column chromatography of the crude metabolite mixture furnished five compounds, 2, 3, 4, 5, and 6. Compound 2 was



* To whom correspondence should be addressed. Tel: (662) 915-1019. Fax: (662) 915-7989. E-mail: amgalalv@yahoo.com. **Table 1.** ¹³C NMR Chemical Shift Assignments of Compounds **2**, **3**, **5**, and 7^a

position	2	3	5	7 ^b
1	78.6, d	76.1, d	78.3, d	79.1, d
2	31.8, t	33.5, t	27.9, t	24.2, t
3	34.7, t	123.3, d	40.2, t	40.2, t
4	143.1, s	133.8, s	72.3, s	71.5, s
5	54.0, d	54.3, d	57.2, d	56.9, d
6	69.9, d	71.6, d	72.6, d	71.8, d
7	56.6, d	59.6, d	57.9, d	57.3, d
8	76.5, d	76.9, d	76.4, d	75.3, d
9	40.4, t	39.4, t	43.2, t	42.4, t
10	43.0, s	41.1, s	41.8, s	40.5, s
11	40.9, d	41.9, d	41.1, d	40.3, d
12	178.9, s	179.6, s	178.6, s	177.6, s
13	14.2, q	15.1, q	14.4, q	13.9, q
14	14.2, q	13.3, q	16.2, q	16.8, q
15	109.0, t	24.0, q	24.7, q	24.2, q
16				21.0, q
17				169.6, s
18	21.3, q		21.8, q	21.3, q
19	171.3, s		170.1, s	170.2, s

 a In CDCl3 at 125 MHz. Carbon multiplicities were determined by DEPT 135° experiments. b At 75 MHz.

previously isolated from the flowers of *Chrysanthemum cinerariaefolium* Visiani (Pyrethrum flower), while **6** was isolated from *Cassinia loevis* R.Br. Compounds **3**, **4**, and **6** were previously synthesized from pyrethrosin (**1**) using BF_3 ·OEt₂ and *p*-toluenesulfonic acid to convert the germacrane ring into the eudesmane skeleton.¹⁰

The positive ion HRESIFTMS spectrum of 5 displayed a molecular ion peak $[M + H]^+$ at m/z 327.1822, suggesting the molecular formula $C_{17}H_{24}O_6$. The IR spectrum of 5 showed a broad absorption band at 3427 cm⁻¹, suggesting the presence of a hydroxyl group. It also showed a strong absorption band at 1767 cm⁻¹, consistent with the existence of a γ -lactone carbonyl functionality. The ¹³C and ¹H NMR data of 5 (Table 1 and experimental) were in agreement with a eudesmanolide skeleton.¹² Inspection of the NMR spectra of 5 revealed close similarity to those of the previously reported compound 6α -acetoxy- 1β , 4α -dihydroxy-5,7 α *H*,8 β *H*-eudes-11-en-8,12-olide (**6**),¹² except for the presence of a methyl group at C-11, resonating at δ 14.4 (d, Me-13) in 5, instead of an exocyclic methylene functionality in 6. The assignment of the relative stereochemistry of the stereocenter C-11 was deduced from NOESY and ROESY data. The β -oriented Me-15 displayed NOESY correlation with the downfield H-11, suggesting a similar stereoorientation. Moreover, the α -oriented acetate methyl at C-6 showed ROESY correlation with Me-13, suggesting α -orientation for Me-13 and supporting the relative stereochemistry assignment of C-13. Therefore, metabolite 5 was established as 6α -acetoxy- 1β , 4α -dihydroxy-5, $7\alpha H$, $8\beta H$ eudesm-11 β ,13-dihydro-8,12-olide.

Metabolites 2, 3, 4, and 6 were identified by comparing their spectral data with literature.^{3,10} All isolated new and known compounds were tested for cytotoxicity, antimicrobial, and antimalarial activities using standard methods described elsewhere.11-13

Metabolite 5 and its 1-O-acetate derivative (7) exhibited antiprotozoal activity against Plasmodium falciparum (D6 clone) with IC₅₀ 0.88 and 0.32 μ g/mL, respectively, without significant toxicity. Compound 7 also displayed a pronounced activity against the chloroquine-resistant strain of P. falciparum (W2 clone) with IC₅₀ 0.38 µg/mL. Compound 2 showed considerable in vitro cytotoxic activity against human epidermoid carcinoma (KB) and against human ovary carcinoma (SK-OV-3) with IC₅₀ < 1.1 and 8.0 μ g/mL, respectively. Pyrethrosin (1) and compound 3 exhibited cytotoxic activity against human malignant melanoma (SK-MEL) with IC₅₀ 4.20 and 7.5 μ g/mL, respectively. Compounds 4 and 7 were active against *Cryptococcus neoformans* with IC₅₀ 35.0 and 25 μ g/mL, respectively, while **1** and **5** displayed anticandidal activity (*Candida albicans* B311) with IC₅₀ 30 and 10 μ g/mL, respectively. Metabolites 2, 3, 4, 5, and 7 are more cytotoxic and antifungal than the parent compound (1).⁴

Experimental Section

General Experimental Procedures. IR spectra were recorded with an ATI Mattson Genesis Series Fourier transform (FT-IR) spectrophotometer. Optical rotations were recorded at ambient temperature using a JASCO DIP-370 digital polarimeter. 1D and 2D NMR spectra were obtained on Bruker Avance DRX 400 and 500 spectrometers. HRESIFTMS was obtained using a Bruker Bioapex FT-MS in ESI mode. For TLC, glass-supported silica gel 60 plates (0.25 mm layer, F_{250} , E. Merck) were used. Silica gel 60 (230-400 mesh, E. Merck) was used for column chromatography. Visualization was accomplished by spraying with p-anisaldehyde spray reagent followed by heating at 110 °C.

Chemicals. Pyrethrosin was kindly supplied by Dr. F. S. El-Feraly, College of Pharmacy, King Saud University, Riyadh, Saudi Arabia. The compound was authenticated by comparing its physical and spectral data with literature values.²

Organisms. Microbial metabolism studies were conducted as previously reported.¹⁴ Twenty-four microbial cultures were obtained from either American Type Culture Collection (ATCC) or National Center for Agricultural Utilization Research (NCAUR) (formerly Northern Regional Research Laboratories, NRRL); organisms were maintained on Sabourad dextrose agar (Oxoid) slants at 4 °C and were used for the preliminary screening. All fermentation experiments were carried out in a medium of the following composition: 10 g of dextrose, 10 mL of glycerol, 5 g of peptone, 5 g of K₂HPO₄, 5 g of NaCl, and 1000 mL of distilled H₂O. The pH was adjusted to 6.0 before autoclaving at 121 °C for 15 min.14

Fermentation Procedures.¹⁴ Cells of microorganisms were transferred from 2 week old slants into 200 mL of sterile culture media per 1 L flask and kept on a gyratory shaker at room temperature, 80 rpm, for 72 h to give stage I culture. Stage I cultures (5 mL) were used as inocula for stage II cultures (50 mL/250 mL flask). After 24 h incubation of stage II cultures, **1** was added as a solution in Me_2CO (5 mg/0.25 mL), per flask. Both substrate and culture controls were run. Each fermentation was sampled by extracting 5 mL of the culture medium with CHCl₃ (5 mL), every 3 and 6 days.

Fermentation of Pyrethrosin (1). Pyrethrosin (1, 410 mg), dissolved in Me₂CO, was added at 10 mL per flask in 41 flasks (250 mL) containing Stage I cultures of Rhizopus nigricans NRRL 1477. Fermentation was stopped after 6 days.

The mixture was filtered, and the fermentation broth was extracted three times with equal volumes of CHCl₃. The combined extracts were dried over anhydrous Na₂SO₄, and the solvent was evaporated to give 425 mg of translucent residue. The crude residue was loaded on a silica gel column (45 g). Elution of the column utilizing a gradient of EtOAc in hexane $(50 \rightarrow 100\%)$ yielded **3** as a yellowish powder (32.0 mg) and **5** (75 mg). The impure 2 (49 mg) was rechromatographed on a silica gel column (7.5 g) and eluted with 40% EtOAc in hexane. This column afforded 2 (17 mg, colorless amorphous solid). Impure 4 was also rechromatographed over a silica gel column (10.0 g). Elution with 50% EtOAc in hexane afforded 4 (18 mg, colorless amorphous solid).

6α-Acetoxy-1β,4α-dihydroxy-5,7α*H*,8β*H*-eudesm-11β,13**dihydro-8,12-olide 5:** pale yellow gum, $[\alpha]_{D}$ +28.3 (*c* 0.06, CHCl₃); IR v_{max} (film) 3427 (OH), 2939, 2876, 1767 (C=O), 1745 (C=O), 1456, 1375, 1251, 1149, 1092, 1030 cm⁻¹; ¹H NMR (500 MHz, CDCl3) δ 1.05 (3H, s, Me-14), 1.17 (3H, d, J = 6.8 Hz, Me-13), 1.21 (3H, s, Me-15), 1.36 (1H, m, H-9'), 1.63 (2H, m, H-2', H-3'), 1.77 (2H, m, H-2, H-3), 1.84 (2H, m, H-5, H-7) 2.15 (3H, s, Me-18), 2.50 (1H, dd, J = 3.8, 3.1 Hz, H-9), 2.62 (1H, septet, J = 6.7 Hz, H-11), 3.10 (1H, s, OH-4), 3.74 (1H, dd, J = 10.3, 4.0 Hz, H-1), 4.07 (1H, ddd, J = 11.7,3.9,3.7 Hz, H-8), 5.47 (1H, t, J = 10.6 Hz, H-6); ¹³C NMR, see Table 1; HRESIFTMS m/z 327.1822 [M + 1]⁺ (calcd for C₁₇H₂₆O₆, 327.1808).

Acetylation of 5. To a solution of 5 (6.5 mg) in pyridine (0.5 mL) was added acetic anhydride (0.5 mL), and the reaction mixture was kept at room temperature for 7 h. The reaction mixture was diluted with CH₂Cl₂ (10 mL) and shaken with 2% aqueous NaHCO₃, followed by shaking with 2% HCl. The CH₂Cl₂ extract was then dried over anhydrous Na₂SO₄ to yield 7 (7.0 mg, 95.4%, colorless amorphous solid, *R*_f 0.21, 1% MeOH in CH₂Cl₂): $[\alpha]$ +50.0 (*c* 0.04, MeOH); IR ν_{max} (film) 3580-3478 (OH), 2955, 1780 (C=O), 1736 (C=O), 1373, 1241, 1154, 1083, 1030, 980 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.12 (3H, s, Me-14), 1.17 (3H, d, J = 6.8 Hz, Me-13), 1.22 (3H, S, Me-15), 1.87 (2H, m, H-5, H-7), 2.06 (3H, s, Me-16), 2.15 (3H, s, Me-18), 2.60 (1H, septet, J = 6.0 Hz, H-11), 3.06 (1H, s, OH-4), 4.02 (1H, ddd, J = 11.6, 3.7, 3.1 Hz, H-8), 4.68 (1H, brd, J = 9.4 Hz), 5.45 (1H, t, J = 10.5 Hz); ¹³C NMR, see Table 1; HRESIFTMS m/z 403.1536 [M + Cl]⁻ (calcd for C₁₉H₂₈O₇, 403.1529).

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References and Notes

- (1) Barton, D. H. R.; De Mayo, P. J. Chem. Soc. 1957, Part V111, 150-158.
- (2) Iriuchijima, S.; Tamura, S. *Agr. Biol. Chem.* **1970**, 204–209.
 (3) Doskotch, R. W.; El-Feraly, F. S.; Hufford, C. D. *Can. J. Chem.* **1971**,
- 49. 2103-2110.
- (4) Abeysekera, B. F.; Abramowski, Z.; Towers, G. H. N. Biochem. Syst. Ecol. 1985, 13, 365–369.
- (5) Picman, A. K.; Towers, G. H. N. Biochem. Syst. Ecol. 1983, 11, 321-
- (6) Picman, A. K. *Biochem. Syst Ecol.* 1984, *12*, 13–18.
 (7) Yoke, M. Y.; Balza, F.; Abeysekera, B. F.; Towers, G. H. N. *Biochem. Syst. Ecol.* 1984, *12*, 285–286.
- (8) Iino, Y.; Tanaka, A.; Yamashita, K. Agr. Biol. Chem. 1972, 2505-2509. (9)Abourashed, E. A.; Clark, A. M.; Hufford, C. D. Curr. Med. Chem.
- 1999, 6, 359-374.
- (10) Cardona, M. L.; Fernandez, I.; Garcia, B.; Pedro, J. R. J. Nat. Prod. 1990, 53, 1042–1045. (11) Borenfreund, E.; Abich, H. B.; Martin-Alguacil, N. Toxic. In Vitro.
- 1988, 2, 1-6 (12)Li, X.; ElSohly, H. N.; Nimrod, A. C.; Clark, A. M. J. Nat. Prod. 1999, 62, 674-677
- El Sayed, K. A.; Dunbar, D. C.; Goins, D. K.; Cordoya, C. R.; Perry, (13)T. L.; Weeson, K. J.; Sanders, S. C.; Janus, S. A.; Humman, M. J. Nat. Toxins **1996**, *5*, 261–285.
- (14) Galal, A. M.; Ibrahim, A. S.; Mossa, J. S.; El-Feraly, F. S. Phytochemistry 1999, 51, 761-765.

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